



**UNIVERSITY OF RUHUNA – FACULTY OF MEDICINE**  
**ALLIED HEALTH SCIENCES DEGREE PROGRAMME**  
**FOURTH BPHARM PART I EXAMINATION – DECEMBER 2016**  
**PH 4134 PHARMACEUTICAL TECHNOLOGY (SEQ)**

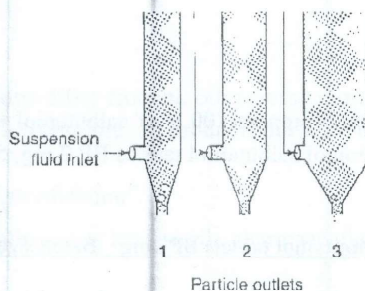
**TIME: THREE HOURS**

**INSTRUCTIONS**

- Answer *all* questions.
- Do not use any correction fluid.
- Answer questions in the booklet/s provided.
- Marks will be deducted for illegible hand writing.

1.

- 1.1 Explain the reasons for granulation. (20 marks)
- 1.2 Explain the three (03) mechanisms of granule formation in wet granulation. (30 marks)
- 1.3 Give a brief account on methods that can be utilized to increase the filtration rate of a fluid. (15 marks)
- 1.4 Following figure show a continuous elutriation process. Describe how this process can efficiently separate particles. (15 marks)



- 1.5 Explain briefly the extrusion and spheronization. (20 marks)

2. Write short notes on followings (Sketch the diagrams wherever necessary).

- 2.1 Shear thinning and shear thickening rheological systems (25 marks)
- 2.2 Negative and positive mixtures (25 marks)
- 2.3 Bed drying and oven drying (25 marks)
- 2.4 Hammer mill (25 marks)

3.

3.1 What is meant by scale of scrutiny? (05 marks)

3.2 If number of particles are 36000, calculate the standard deviation of active ingredient in the mixture. (10 marks)

3.3 Using necessary equations show how changes in active ingredient would affect the variation of the active ingredients. (15 marks)

3.4 The relative viscosity ( $\eta_r$ ) of a solution is given by the equation given below,

$$\eta_r = \frac{\eta}{\eta_0}$$

Where viscosity of solution ( $\eta$ ) and viscosity of the solvent ( $\eta_0$ ),

Using relative viscosity and specific viscosity, derive an equation to calculate the viscosity of a colloidal suspension. (10 marks)

3.5 Briefly describe why some hydrates show higher solubility where as some show lower solubility characteristics. (20 marks)

3.6 "Formulation of drugs using amorphous materials is a prominent solution for poorly soluble drugs". Do you agree with this statement? Justify your answer. (20 marks)

3.8 Electricity and steam can be used as heating media. Discuss briefly the pros and cons of both of these media. (20 marks)

4. Assume that you are requested to prepare 100 g of salbutamol tablet BP 2 mg batch. The master formula for the production of salbutamol tablets BP 2 mg, 500 g batch is given below.

**Formula for salbutamol tablets BP 2mg Batch size 500g**

Salbutamol sulphate BP	006.020 g
Lactose BP	334.780 g
Maize starch BP	115.000 g
Maize starch BP	030.000 g
Purified talc BP	025.000 g
Magnesium stearate BP	022.500 g
Purified water 60°C	075.000 mL
Purified water 100°C	130.000 mL

4.1 Calculate the working formula. (10 marks)

4.2 State the purpose of using following ingredients in above production. (15 marks)

4.2.1 Salbutamol sulphate BP

4.2.2 Lactose BP

4.2.3 Purified talc

4.3 During the production process of salbutamol tablets, maize starch mucilage is prepared by mixing 30 g of maize starch in 75 mL of purified water at 60 °C and then transferring that in to 130 mL of boiling water.

4.3.1 State the purpose of adding maize starch mucilage in this salbutamol tablet production process. (05 marks)

4.3.2 Name three (03) types of granulators that can be used for above process. (10 marks)

4.4 Name steps that used to prepare the powder mixture mentioned in above 4.1 to be compressed. (20 marks)

4.5 Explain the “tableting procedure” of above 4.3.2 prepared granules using single punch tablet press. (40 marks)

5.

5.1 Briefly describe the rotary die process used to prepare soft gelatin capsules. (30 marks)

5.2 State two (02) methods to modify the release profile of capsules. (10 marks)

5.3 Describe the importance of conducting pilot plant studies. (20 marks)

5.4 Briefly describe the use of “stem cell technology” in cosmetology. (25 marks)

5.5 In industries, safety devices are used to prevent, reduce or eliminate hazards which cause harmful effects. Sate five (05) such devices available in industries. (15 marks)

6. Parenteral dosage forms differ from all other dosage forms, because they are injected directly in to the body tissues. Therefore, all products must be sterile.

6.1 Define the term “sterilization”. (05 marks)

6.2 List five (05) requirements for a sterile pharmaceutical product manufacturing plant. (15 marks)

6.3 State the reason for adding preservatives to the parenteral products even though those products are sterilized. (05 marks)

6.4 Using labeled diagrams, briefly explain the reason for using vertical laminar airflow hoods instead of horizontal airflow hoods in the preparation of chemotherapy drugs. (20 marks)

6.5 Briefly describe the “barrier isolator technology” used in parenteral production. (25 marks)

6.6 For the production of parenteral products, water for injection (WFI) produced by distillation.

6.6.1 Sketch a cross section of a multiple effect still. (10 marks)

6.6.2 Briefly describe the production process of WFI and mark the material flow path on the drawn sketch. (20 marks)