



# University of Ruhuna

## Bachelor of Science Degree in Fisheries and Marine Sciences

### Level I (Semester II) Examination

December 2015

Subject: Mathematics

Course Unit: FSC1b50

Time: Two (02) Hours

Answer ALL Questions. Each Question carries equal marks.

1. (a) Find  $a$  and  $b$  such that

$$a + ib = \frac{1 - z}{1 + z},$$

where  $z = 1 + 2i$  and  $i$  is the imaginary unit.

- (b) Using the binomial theorem, show that

$$\left(x - \frac{1}{x}\right)^4 = x^4 - 4x^2 + 6 - \frac{4}{x^2} + \frac{1}{x^4}.$$

- (c) Solve the following equations for  $x$ :

(i)  $e^{2x} - 3e^x + 2 = 0$

(ii)  $\log_{10}(x+8) + \log_{10}(x-1) = 1$

- (d) (i) Write down the expressions for  $\cos(\alpha + \beta)$  and  $\cos(\alpha - \beta)$ .

Prove that

$$\cos \alpha \cos \beta = \frac{1}{2} [\cos(\alpha + \beta) + \cos(\alpha - \beta)].$$

Hence show that

$$\cos 37.5^\circ \cos 7.5^\circ = \frac{\sqrt{2} + \sqrt{3}}{4}.$$

- (ii) Verify the following trigonometric identity:

$$\frac{\sin(\alpha + \beta)}{\sin(\alpha - \beta)} = \frac{\tan \alpha + \tan \beta}{\tan \alpha - \tan \beta}.$$

2. (a) Find the following limits:

(i)  $\lim_{h \rightarrow 1} \frac{h^4 - 1}{h^3 - h}$

(ii)  $\lim_{t \rightarrow 3} \frac{t^2 - 8t + 15}{t^2 + 4t - 21}$

- (b) Differentiate the following functions with respect to  $x$ :

(i)  $y = e^{3x} \sin 3x$

(ii)  $y = \frac{x^2 + x + 1}{1 - x}$

- (c) Consider the curve given by  $y = x^2 + bx + c$ , where  $b$  and  $c$  are constants. Find the values of  $b$  and  $c$  such that the straight line  $y = 2x$  is tangent to the curve at the point  $(2, 4)$ .
- (d) Find the turning points of the curve given by the function  $f(x) = 4x^3 + 15x^2 - 18x + 6$  and identify them as maxima or minima using the second derivative  $f''(x)$ .

3. (a) Consider the function given by

$$f(x, y) = \sqrt[3]{x^2 + y}$$

Show that the total differential of  $f$  at the point  $(2, 4)$  is given by

$$df = \frac{1}{12} [4dx + dy]$$

- (b) Show that the function  $f(x, y) = 9x^3y + 8x^2y^2 - 6xy^3$  is homogeneous and satisfies the Euler's theorem.
- (c) Use integration by parts to evaluate

$$\int x \cos x dx.$$

Using integration by parts and the above result, show that

$$\int x^2 \sin x dx = -x^2 \cos x + 2x \sin x + 2 \cos x + C,$$

where  $C$  is an arbitrary constant.

- (d) Show that

$$\int_1^2 \frac{3t}{t^2 + 4} dt = \frac{3}{2} \ln \frac{8}{5}.$$

4. (a) Find the general solution of the differential equation

$$\frac{1}{y} \frac{dy}{dx} = \frac{x}{x^2 + 1}$$

using the technique of separation of variables.

- (b) Test the differential equation

$$(2xy + y^3 \cos x) dx + (x^2 + 3y^2 \sin x) dy = 0$$

for exactness. If it is exact, then find its solution.

- (c) The following table contains the number of championships the clubs A, B, C, D, E, F and G have won over a given period.

Club	A	B	C	D	E	F	G
No. of championships won	9	5	7	$x$	3	$x+1$	4

If these clubs won 5 championships over the given period of time on average, how many championships did the clubs D and F win over the given period?

Also find the mode, median, range, mean deviation, variance and standard deviation of the number of championships won by these clubs over the given period.

# The Periodic Table of the Elements

	1											13	14	15	16	17	18	
1	1 <b>H</b> Hydrogen 1.00794																2 <b>He</b> Helium 4.003	
2	3 <b>Li</b> Lithium 6.941	4 <b>Be</b> Beryllium 9.012182											5 <b>B</b> Boron 10.811	6 <b>C</b> Carbon 12.0107	7 <b>N</b> Nitrogen 14.00674	8 <b>O</b> Oxygen 15.9994	9 <b>F</b> Fluorine 18.9984032	10 <b>Ne</b> Neon 20.1797
3	11 <b>Na</b> Sodium 22.989770	12 <b>Mg</b> Magnesium 24.3050	3	4	5	6	7	8	9	10	11	12	13 <b>Al</b> Aluminum 26.981538	14 <b>Si</b> Silicon 28.0855	15 <b>P</b> Phosphorus 30.973761	16 <b>S</b> Sulfur 32.066	17 <b>Cl</b> Chlorine 35.4527	18 <b>Ar</b> Argon 39.948
4	19 <b>K</b> Potassium 39.0983	20 <b>Ca</b> Calcium 40.078	21 <b>Sc</b> Scandium 44.955910	22 <b>Ti</b> Titanium 47.867	23 <b>V</b> Vanadium 50.9415	24 <b>Cr</b> Chromium 51.9961	25 <b>Mn</b> Manganese 54.938049	26 <b>Fe</b> Iron 55.845	27 <b>Co</b> Cobalt 58.933200	28 <b>Ni</b> Nickel 58.6934	29 <b>Cu</b> Copper 63.546	30 <b>Zn</b> Zinc 65.39	31 <b>Ga</b> Gallium 69.723	32 <b>Ge</b> Germanium 72.61	33 <b>As</b> Arsenic 74.92160	34 <b>Se</b> Selenium 78.96	35 <b>Br</b> Bromine 79.904	36 <b>Kr</b> Krypton 83.80
5	37 <b>Rb</b> Rubidium 85.4678	38 <b>Sr</b> Strontium 87.62	39 <b>Y</b> Yttrium 88.90585	40 <b>Zr</b> Zirconium 91.224	41 <b>Nb</b> Niobium 92.90638	42 <b>Mo</b> Molybdenum 95.94	43 <b>Tc</b> Technetium (98)	44 <b>Ru</b> Ruthenium 101.07	45 <b>Rh</b> Rhodium 102.90550	46 <b>Pd</b> Palladium 106.42	47 <b>Ag</b> Silver 107.8682	48 <b>Cd</b> Cadmium 112.411	49 <b>In</b> Indium 114.818	50 <b>Sn</b> Tin 118.710	51 <b>Sb</b> Antimony 121.760	52 <b>Te</b> Tellurium 127.60	53 <b>I</b> Iodine 126.90447	54 <b>Xe</b> Xenon 131.29
6	55 <b>Cs</b> Cesium 132.90545	56 <b>Ba</b> Barium 137.327	57 <b>La</b> Lanthanum 138.9055	72 <b>Hf</b> Hafnium 178.49	73 <b>Ta</b> Tantalum 180.9479	74 <b>W</b> Tungsten 183.84	75 <b>Re</b> Rhenium 186.207	76 <b>Os</b> Osmium 190.23	77 <b>Ir</b> Iridium 192.217	78 <b>Pt</b> Platinum 195.078	79 <b>Au</b> Gold 196.96655	80 <b>Hg</b> Mercury 200.59	81 <b>Tl</b> Thallium 204.3833	82 <b>Pb</b> Lead 207.2	83 <b>Bi</b> Bismuth 208.98038	84 <b>Po</b> Polonium (209)	85 <b>At</b> Astatine (210)	86 <b>Rn</b> Radon (222)
7	87 <b>Fr</b> Francium (223)	88 <b>Ra</b> Radium (226)	89 <b>Ac</b> Actinium (227)	104 <b>Rf</b> Rutherfordium (261)	105 <b>Db</b> Dubnium (262)	106 <b>Sg</b> Seaborgium (263)	107 <b>Bh</b> Bohrium (262)	108 <b>Hs</b> Hassium (265)	109 <b>Mt</b> Meitnerium (266)	110 (269)	111 (272)	112 (277)	113	114				

Lanthanide series

	58 <b>Ce</b> Cerium 140.116	59 <b>Pr</b> Praseodymium 140.90765	60 <b>Nd</b> Neodymium 144.24	61 <b>Pm</b> Promethium (145)	62 <b>Sm</b> Samarium 150.36	63 <b>Eu</b> Europium 151.964	64 <b>Gd</b> Gadolinium 157.25	65 <b>Tb</b> Terbium 158.92534	66 <b>Dy</b> Dysprosium 162.50	67 <b>Ho</b> Holmium 164.93032	68 <b>Er</b> Erbium 167.26	69 <b>Tm</b> Thulium 168.93421	70 <b>Yb</b> Ytterbium 173.04	71 <b>Lu</b> Lutetium 174.967
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Actinide series

	90 <b>Th</b> Thorium 232.0381	91 <b>Pa</b> Protactinium 231.03588	92 <b>U</b> Uranium 238.0289	93 <b>Np</b> Neptunium (237)	94 <b>Pu</b> Plutonium (244)	95 <b>Am</b> Americium (243)	96 <b>Cm</b> Curium (247)	97 <b>Bk</b> Berkelium (247)	98 <b>Cf</b> Californium (251)	99 <b>Es</b> Einsteinium (252)	100 <b>Fm</b> Fermium (257)	101 <b>Md</b> Mendelevium (258)	102 <b>No</b> Nobelium (259)	103 <b>Lr</b> Lawrencium (262)
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